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The solution is saturated. b. All 15 g of KClO_3 will dissolve; the solution is unsaturated. 3. a. 250. g b. 50. g c. KI 4. a. 1.027 kJ/g b. The temperature of the system will increase because the enthalpy of solution is negative,

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indicating that the reaction is exothermic, giving off energy as heat and warming up the system. SECTION 3 SHORT ...

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homogeneous mixtures
of two or more
substances in a single
phase. Soluble
describes a substance
as capable of being
dissolved.

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