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E1-26 The room volume is $(21 \text{ m} \times 13 \text{ m} \times 12 \text{ m}) = 3276 \text{ m}^3$. The mass contained in the room is $(3276 \text{ m}^3)(1.21 \text{ kg/m}^3) = 3964 \text{ kg}$. E1-27 One mole of sugar cubes would have a volume of $N_A \times 1.0 \text{ cm}^3$, where N_A is the Avogadro constant. Since the volume of a cube is equal to the length cubed, $V = l^3$, then $l = \sqrt[3]{V} = \sqrt[3]{N_A \text{ cm}^3} = 1.0 \times 10^8 \text{ cm}$. E1-28 The number of seconds in a week is

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